EREMOPHILANOLIDES, EUDESMANOLIDES, GUAIANOLIDES AND OTHER CONSTITUENTS FROM ONDETIA LINEARIS

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Abstract—The aerial parts of *Ondetia linearis* afforded in addition to known compounds, 13 eremophilanolides, six eudesmanolides, one pseudoguaianolide and two norsesquiterpenes. The structures were elucidated by high field NMR techniques. The chemotaxonomic aspects are discussed briefly.

INTRODUCTION

The monotypic South African genus Ondetia is placed in the second group of the subtribe Inulinae (Compositae, tribe Inuleae) together with Geigeria [1]. While the chemistry of several species of the latter genus has been studied [2-6] nothing was known on the constituents of Ondetia linearis and we therefore studied this South West African species. The results are discussed in this paper.

RESULTS AND DISCUSSION

The extract of the aerial parts gave by careful chromatographic separation techniques the eudesmanolides 1 [7], 2-4, 5 [8], 6-8 and 9 [9], the eremophilanolides 10-16 and 18-23, the noreremophilanes 24 and 25, the cyclopropenone 26 [10], the acid 27, the guaianolides 28 [11] and 29 [12], the pseudoguaianolides 31 [13, 14] as well as the corresponding alcohol 30, the geigeriolides 32 and 33 [6], the lignane derivative 34 [6], vomifoliol, geranylisobutyrate and selina-4,11-diene.

The structures of 2 and 3 followed from their 1H NMR spectra (Table 1) which were similar to that of 1 [7]. However, the signals of methyl doublets and the absence of methylene proton signals required the presence of 11,13-dihydro derivatives. The observed coupling $J_{7,11}$ did not allow a direct assignment of the stereochemistry at C-11. Boranate reduction of 1 afforded a 1:5 mixture of 2 and 3. Inspection of a Dreiding model clearly showed that this result was only in agreement with the proposed configurations at C-11 as the attack of boranate from the α -face was favoured.

The ¹H NMR spectrum of 4 and of the corresponding acetate obtained by mild acetylation (4Ac) (Table 1) indicated that again an eudesmanolide was present. Spin decoupling allowed the assignment of all signals. The obtained sequences required a Δ^3 -double bond and a hydroxy group at C-2. The coupling of H-2 indicated a 2α -hydroxy group. Accordingly, the spectrum was in part close to that of the corresponding costic acid derivative [15]. The ¹H NMR spectra of 6 and 7 (Table 1) were similar to that of 2α -hydroxyalantolactone (5) [8]. Again the observed methyl doublets showed that the corre-

sponding dihydro derivatives isomeric at C-11 may be present. Boranate reduction of 5 gave only the isomer 6. Therefore, this isomer had an 11β -methyl group. In agreement with this in the case of the lactone 7 the observed NOE's indicated the 11α -methyl configuration. Thus effects between H-13, H-7 (5%) and H-8 (4%), between H-15, H-2 (4%), H-6 β (8%) and H-9 β (4%) as well as between H-8 and H-7 (8%) were observed.

The spectrum of **8** (Table 1) was similar to that of **6**. However, the methyl doublet was replaced by a singlet at δ 1.44. In agreement with the molecular formula $(C_{15}H_{22}O_4)$ a 11-hydroxy derivative of **6** was proposed. The configuration at C-11 was deduced from the chemical shift of H-8 which was shifted down field by 0.32 ppm if compared with the shift of H-8 in the spectrum of **6**.

The structure of 10 also followed from its 1 H NMR spectrum (Table 2) which was in part similar to that of the corresponding 1β , 10β -epoxyfuroeremophilane [16]. However, the signals of H-6 to H-9 and H-13 showed that a methylene lactone was present. The configuration at C-1 was assigned from the splitting of H-1 which was a triplet in the α -epoxides while in the β -epoxides a doublet was visible. The configuration at C-8 followed from the splitting of H-8 which was identical with that of H-8 in xanthadiene [17].

The ¹H NMR spectrum of 11 (Table 2) indicated that this lactone was a 11,13-dihydro derivative of compound 10. Boranate reduction of 10 afforded only one dihydro compound which was identical with the natural product. Inspection of a model indicated that this observation only agreed with the presence of an 11β -methyl derivative.

The structure of 12 followed from its 1H NMR spectrum (Table 2) which was in part very similar to that of compound 10. The molecular formula ($C_{15}H_{20}O_4$) showed that an additional oxygen function was present. The IR spectrum required a hydroxy group and a 1H NMR signal at $\delta 4.04$ was coupled with a doublet at $\delta 3.22$ which was due to an epoxide proton. Inspection of a model indicated that the couplings of H-2 best agreed with a 2α -hydroxy derivative.

The spectra of 13 and 14 (Table 2) required the presence of the corresponding 11,13-dihydro derivatives of compound 12. In the case of compound 14 the observed

NOE's established the proposed stereochemistry at all chiral centres. Especially important were effects between H-14, H-2, H-3 and H-9 β , between H-7, H-8 and H-11 as well as between H-11, H-7 and H-8. The couplings $J_{7,11}$ in the isomer 13 was 13 Hz which was in good agreement with an 11α -methyl lactone.

All signals in the spectrum of 15 (Table 2) could be assigned by spin decoupling and the stereochemistry was determined by the observed NOE's in deuteriobenzene between H-14, H-3 (7%), H-9 β (4%) and H-11 (3%), between H-7 and H-8 (7%) as well as between H-1, H-2 α (7%) and H-9 α (6%).

The molecular formula of 16 indicated that this lactone had two hydroxy groups and the 1H NMR spectrum (Table 2) agreed with this assumption. Thus the signals of H-1 and H-9 were double doublets indicating that no hydrogen was located at C-10. An NOE between H-1, H- 2α (7%) and H- 9α (6%) and between H-14 and H- 9β (4%) supported the proposed stereochemistry which was established by acid catalysed hydrolysis of the epoxide 10 which gave a diol identical with the isolated lactone

The ¹H NMR spectrum of 19 (Table 2) differed very slightly from that of 16, only H- 2α was shifted downfield. As the couplings in the spectra of 16 and 19 were nearly

identical the presence of an isomeric diol could be excluded. Therefore a chlorohydrine was proposed with a chlorine at C-10. This would explain the downfield shift of H-2 α as the effect of an axial chloro is stronger than that of a hydroxy group This was established by reaction of the lactone 10 with hydrochloric acid in chloroform which gave the chlorohydrine 19 as the only product, which therefore may be an artifact

The ¹H NMR spectral data of 18 (Table 2) showed that we were dealing with a 11,13-dihydro derivative of 16. Again the assignment of the configuration followed from the result of boranate reduction of 16 which only gave the isomer 17, due to the preferred α -attack of boranate The product 17 was not identical with 18.

The ¹H NMR spectrum of **20** (Table 3) showed a broadened singlet at δ 5.91 Spin decoupling indicated that it was due to H-1 as an allylic coupling with H-9 was observed. The presence of a keto group at C-2 followed from the down field shifts of H-3 which were assigned by irradiation of H-4 The remaining signals were similar to those of the other eremophilanolides

The 1 H NMR spectral data of 21 (Table 3) showed that again an eremophilanolide must be present. Spin decoupling indicated that the low field signals at $\delta 5.55$ and 7.75

4Ac, 6Ac, 8Ac and 30Ac are the corresponding acetates

were due to H-1 and H-2. As the signals of H-9 only were coupled with H-8 a hydroxy group was at C-10. The configuration followed from the chemical shifts of H-14 and H-15 which are not influenced by the hydroxy group as in many other eremophilanes with a *trans*-ring junction.

The ¹H NMR spectrum of 22 (Table 3) was very similar to that of 21. However, the signals of the olefinic protons (H-1 and H-2) were replaced by a doublet at δ 2.92 and a double doublet at 3.37 due to epoxide protons. Furthermore, a narrowly split doublet at δ 2.84 was visible which showed a *W*-coupling with H-9 β . Accordingly, a 10α -hydroxy group was present. The IR band at 3530 cm^{-1} indicated a hydrogen bond. Therefore a 1α , 2α -epoxide was present in agreement with the triplet splitting of H-2.

The ¹H NMR spectrum of 23 in deuteriobenzene (Table 3) clearly showed that a glaucolide-like lactone was present. A pair of broadened double doublets were due to H-13 as they were coupled with a hydroxy triplet at δ 2.01 and sharpened on irradiation of H-8 and H-6. Furthermore the H-6 signals showed only geminal couplings while a doublet at δ 2.40 was due to H-1 as followed from spin decoupling. This signal was at δ 3.08 in CDCl₃. Therefore a 1 β ,10 β -expoxide was present. Most probably lactone 23 was formed biogenetically by oxidation of 10 followed by allylic rearrangement.

The molecular formulae of 24 and 25 indicated that they were norsesquiterpenes. The ^{1}H NMR data of 24 (Table 3) showed that a conjugated ketone was present ($\delta 6.83$ and 5.99 d). Furthermore a triplet of triplets at

Н	2	3	4	4Ac	6	6Ac	7	8
1α	*	1.50 m	*	1 34 dd	1 08 t	1 17 t	1 08 t	1 09 t
1β	*	1 60 m	*	1 99 m	1 95 br d	1.95 br d	1 95 <i>br d</i>	1 95 br d
2 2'	1 86 m 1 71 m	1.78 m 1 66 m	} 4.30 br t	5.38 br t	} 4.20 tt	} 5 29 tt	} 4 20 tt	4 19 tt
3α 3β	3 90 brs	3 94 br s	} 5 46 br s	} 5 41 brs	1.47 dt 1 91 br d	1.57 dt 1 91 br d	1 46 dt 1 91 br d	1 47 dt 1 91 br d
4		_		´—	1 68 br dq	2 70 br dq	2 63 br dq	2 67 br dq
5	-		2.01 br d	2 05 br d		-		
6α	171 m	1 70 m	1 95 m	1.99 d	15251	$\{526d\}$	$\} 2.25 d$	} 5 22 d
6β	2 67 dd	2 44 dd	1 25 m	1.23 dd	5 25 d	3 20 a	} 2.23 a	} 3 22 u
7	2 17 dddd	2 35 dddd	3 02 br tt	3 03 br tt	3 08 ddd	3.07 ddd	2 68 br dd	2.93 dd
8	4 55 ddd	4.45 ddd	4 56 dt	4 52 dt	4 70 ddd	4 73 ddd	4 87 ddd	5 05 ddd
9α	1 64 dd	1 54 dd	1 47 dd	1 53 dd	1 55 dd	1 55 dd	1 54 dd	1 56 dd
9β	2 02 dd	2 24 dd	2 19 br d	2 20 br d	2 19 dd	2 19 dd	2 17 dd	2.20 dd
11	2 42 dq	2 81 dq		_	2.90 dq	2.90 dq	2.44 br q	
13	} 1 30 d	123 d	6 15 br s	6.16 br s	1.23 d	} 1 23 d	1.36 d	} 1 44 s
13'	J)	5 62 br s	5.63 br s))))
14	1.10 s	1 09 s	0 94 s	0.99 s	1 27 s	1 32 s	1 28 s	1 23 s
15	1 79 br s	1 80 br s	1 67 br s	1 69 br s	1 17 d	1 22 d	1 18 d	1 15 d
OAc				205s		2 03 s		

Table 1 ¹H NMR spectral data of compounds 2-4, 4Ac, 6, 6Ac, 7 and 8 (CDCl₃, 400 MHz, δ-values)

J [Hz]: Compounds 2 and 3 2α ,3 = 2β ,3 ~ 2, 6α ,6 β = 14, 11,13 = 7 (compound 2. 6α ,7 = 6β ,7 = 7,8 ~ 7,7,11 = 4, 8,9 α = 4, 8,9 β = 7, 9α ,9 β = 15, compound 3 6α ,7 = 6; 6β ,7 = 12, 7,8 = 4; 7,11 = 7; 8,9 α = 4, 8,9 β = 2; 9α ,9 β = 15.5), compounds 4 and 4Ac 1α ,1 β = 1α ,2 = 12, 1β ,2 ~ 7, 2,3 ~ 2, 5,6 β = 6α ,6 β ~ 12; 6α ,7 = 6β ,7 = 7, 7,8 = 8 9 α = 5; 9α ,9 β = 15, compounds 6 and 6Ac 1α ,1 β = 1α ,2 = 2, 3α = 12, 1β ,2 = 2,3 β = 4, 1β ,3 β = 3β ,4 = 1.5; 3α ,4 = 6; 4,15 = 11,13 = 7; 6,7 = 3, 7,8 = 5 5, 7,11 = 8, 8,9 α = 2 5, 8,9 β = 3, 9α ,9 β = 15, compound 7 1α ,1 β = 1α ,2 = 2,3 α = 12; 1β ,2 = 2,3 β = 4, 1β ,3 β = 3β ,4 = 1 5, 4,15 = 11,13 = 7, 6,7 = 3, 7,8 = 6, 7,11 ~ 1, 8,9 α = 2 5, 8,9 β = 3, 9α ,9 β = 14, compound 8 1α ,1 β = 1α ,2 = 2,3 α = 12, 1β ,2 = 2,3 β = 4, 3α ,4 = 6, 4.15 = 7; 6,7 = 3 5, 7,8 = 6, 8,9 α = 2 5, 8,9 β = 3, 9α ,9 β = 15

 $\delta4.14$ required an equatorial hydroxy group with two neighbouring methylene groups. Accordingly, a 2α -hydroxy derivative of a noreremophilane was very likely. Spin decoupling supported this proposal. However, the relative position of the keto group had to be determined. This, and the stereochemistry, were established by the observed NOE's. Clear effects were obtained between H-14, H-4 (7%), H-6 (8%) and H-9 β (4%) as well as between H-15, H-2 (4%) and H-6 (5%). Thus a Δ^6 -double bond had to be assumed. Inspection of a model showed that the NMR data required a *cis*-ring junction.

The ¹H NMR spectrum of **25** (Table 3) indicated that again a conjugated ketone must be present. A doublet at $\delta 6.18$ showed an allylic coupling with H-1. Accordingly, the double bond was between C-9 and C-10. The couplings of H-1 required an α -hydroxy group and the remaining signals showed that again a noreremophilane was present.

The ¹H NMR spectrum of 27 (see Experimental) showed that an eremophilanic acid was present Spin decoupling allowed the assignment of all signals though some were overlapping multiplets. The isolation of 27 is of interest as it may be the precursor of the main constituent 10 which itself is the precursor of nearly all the other eremophilanolides (11–18).

The structure of the pseudoguaianolide 30 followed from the ¹H NMR data and from those of the corresponding acetate 30Ac (Table 3) which were in part similar to those of 31 [13]. Spin decoupling and the

observed NOE's of 30Ac established the proposed structure. Clear NOE's were obtained between H-4, H-1 (3%) and H-6 α (5%), between H-8 and H-10 (7%) as well as between H-15 and H-8 (4%). The corresponding 2α -acetoxy derivative was isolated from a Geigeria species [6]. The ¹H NMR spectrum of the latter was similar to that of 30.

The chemistry of Ondetia supports in part the proposed close relationship to Geigeria [1] where a variety of sesquiterpene lactones have been isolated. In particular the co-occurrence of the rare lactones of type 32 [6] is remarkable. But eudesmanolides and eremophilanolides are present also in Geigeria aspera [6] which contains the guaianolide 29 while further guaianolides and some pseudoguaianolides related to 30 and 31 are reported from other Geigeria species [2-6]. However, most remarkable is the extreme variety of lactones in Ondetia The lignane 34 was also present in Geigeria while the rare cyclopropenone 26 has been isolated only from members of the Inuleae from a Telekia species [10] Noreremophilanes of type 24 and 25 are not common They are probably the result of oxidative degradation as this species seems to be very rich in oxidizing enzymes. The proposed relationship of Geigeria and Ondetia to Calostephane and related genera so far is not supported by the chemistry as the latter genus can be characterized by a variety of eudesman- $12,6\beta$ -olides [18] However, the lactone 33 was isolated [19] from the related genus [1] Antiphiona

^{*}Obscured multiplets.

Table 2. ¹H NMR spectral data of compounds 10-19 and 24 (400 MHz, CDCl₃, δ-values)

						J		`	, , , , , , , , , , , , , , , , , , ,			
н	10	11	12	13	14	15	16	C_bD_b	11	18	19	7
-	2 94 br d	3.04 t	3.22 d	3.24 d	3.25 d	3.01 d	3.55 dd	3.14 dd	3.55 t	3.58 t	3.89 dd	{ 1.90 t 1.59 ddd
2a 28	1.92 m 1.27 m	2.02 m 1.87 m	4.04 br s 4.05 br s	$\left. \left. \right. \right. \right\}$ 4.05 br s	} 4.08 ddd	2.44 dddd 1.89 m	2.03 m 1.60 br d	1.87 dddd 1.44 dddd	2.05 m 1.57 m	2.05 m 1.60 m	2.34 dddd 1.88 m	} 4.14 tt
3¢,	1.18 m	1.17 m	×	1.36 br d	1.50 ddd	12.46.444	1.26 m	1.12 dddd	1.29 br d	1.30 m	1.34 br d	2.05 dt
3β	1.92 m	1.48 dddd	×	1.53 m	1.60 ddd	} 3.40 uuu	1.57 m	1.59 dddd	1.60 m	1.60 m	1.70 dddd	1.84 dddd
4	1.70 ddg	1.99 m	1.56 m	1.77 m	2.02 ddg	1.58 m	1.74 ddq	1.69 m	1.67 m	1.74 ddq	1.76 m	2.17 br dq
6α 6 <i>β</i>	2.20 dd 1.83 dd	1.77 dd 0.95 t	2.27 dd 1.83 dd	1.95 br d 1.71 m	1.75 dd 0.98 t	1.89 m 1.73 dd	} 2.00 m	1.81 dd 1.59 br d	} 1.60 m	1.87 dd 1.62 dd	2.02 dd 1.99 d	} 6.83 d
. ~	3.33 ddddd	2.68 ddd	3.36 ddddd	2.90 br tt	2.65 dddd	2.37 m	3.27 dddd	2.84 dddd	2.93 ddddd	2.34 br ddd	3.29 dddd	5.99 brd
~	4.87 ddd	4.72 ddd	4.90 ddd	4.77 ddd	4.71 ddd	4.78 ddd	5.04 ddd	4.86 ddd	4.71 ddd	4.92 ddd	4.99 ddd	I
δ	1.64 dd	1.83 dd	175 dd	171 dd	1.98 dd	1.65 dd	7 04	1.94 dd	1.80 dd	1.91 dd	2.05 dd	2.96 br d
θ6	2.01 dd	2.17 dd	2.05 dd	2.20 dd	2.20 dd	2.15 dd	M / €.1 {	1.71 dd	2.73 dd	2.19 dd	2.17 dd	2.00 br d
Ξ	1	2.82 dq	1	2.71 dq	2.83 dq	2.68 dq	1	1	2.77 dq	2.75 dq	l	1
13	6.37 d	1.16 d	6.40 d	1.29 d	1148	1.28 d	6.34 d	6.30 d	} 1.19 d	1.25 d	6.37 d	1
13′	5.65 d	_	567 d	_		_	5.62 d	5.09 d	<u></u>	ſ	5.63 d	ı
14	0.93 s	0.97 s	0.91 s	1.04 s	0.89 s	1.10 s	1.01 s	0.97 s	1.05 s	1.18 s	1.12 s	1.17 s
15	0.78 d	0.79 d	0.83 d	0.79 d	P 08:0	0.93 d	0.87 d	0.78 d	0.83 d	0.83 d	p 06:0	1.14 d

 $J [Hz] \cdot 4,15 = 7, 6\alpha,6\beta = 9\alpha,9\beta = 14; compound 10 \cdot 1,2 = 4; 3\alpha,4 = 4; 3\beta,4 = 7, 6\alpha,7 = 1.5, 6\beta,7 = 7.5; 7,8 = 7, 7,13 = 35; 7,13' = 3; 8,9\alpha = 6.5, 8,9\beta = 10.5, compound 12 \cdot 1,2 = 4; 3\alpha,3\beta = 15; 6\alpha,7 = 6\beta,7 = 7,8 = 7,7,11 = 13; 8,9\alpha = 7; 8,9\beta = 11, compound 14: 1,2 = 2,3\alpha = 3; 2,3\beta = 8; 3\alpha,4 = 6; 3\beta,4 = 11; 6\alpha,7 = 4.5, 6\beta,7 = 7,8 = 7,7,11 = 13; 8,9\alpha = 7; 8,9\beta = 11; compound 14: 1,2 = 2,3\alpha = 3; 2,3\beta = 8; 3\alpha,4 = 6; 3\beta,4 = 11; 6\alpha,7 = 4.5, 6\beta,7 = 7,8 = 7,7,11 = 13; 8,9\alpha = 6; 8,9\beta = 11; compound 15: 1,2\alpha = 4, 2\alpha,2\beta = 15; 2\alpha,3 = 2\beta,3 = 8; 3,4 = 10; 6\beta,7 = 7,8 = 7,7,11 = 13; 8,9\alpha = 6; 8,9\beta = 11; compound 15: 1,2\alpha = 4, 2\alpha,2\beta = 15; 2\alpha,3 = 2\beta,3 = 8; 3,4 = 10; 6\beta,7 = 7,8 = 6; 7,11 = 13; 8,9\alpha = 6; 8,9\beta = 11; compound 15: 1,2\alpha = 4, 2\alpha,2\beta = 15; 2\alpha,3 = 8; 3,4 = 10; 6\beta,7 = 7,8 = 7,7,11 = 13; 8,9\alpha = 6; 8,9\beta = 11; compound 15: 1,2\alpha = 4, 2\alpha,2\beta = 15; 2\alpha,3 = 8; 3,4 = 10; 6\beta,7 = 7,8 = 7,7,11 = 13; 8,9\alpha = 6; 8,9\beta = 11; compound 15: 1,2\alpha = 4, 2\alpha,2\beta = 15; 2\alpha,3 = 8; 3,4 = 10; 6\beta,7 = 7,8 = 7,7,11 = 13; 8,9\alpha = 6; 8,9\beta = 11; compound 15: 1,2\alpha = 4, 2\alpha,2\beta = 15; 2\alpha,3 = 8; 3,4 = 10; 6\beta,7 = 7,8 = 7,7,11 = 13; 8,9\alpha = 6; 8,9\beta = 11; compound 15: 1,2\alpha = 4, 2\alpha,2\beta = 15; 2\alpha,3 = 8; 3,4 = 10; 6\beta,7 = 7,8 = 7,7,11 = 13; 8,9\alpha = 6; 8,9\beta = 11; compound 15: 1,2\alpha = 4, 2\alpha,2\beta = 15; 2\alpha,3 = 8; 3,4 = 10; 6\beta,7 = 7,8 = 7,7,11 = 13; 8,9\alpha = 6; 8,9\beta = 11; compound 15: 1,2\alpha = 4, 2\alpha,2\beta = 15; 2\alpha,3 = 8; 3,4 = 10; 6\beta,7 = 7,8 = 7,7,11 = 13; 8,9\alpha = 6; 8,9\beta = 11; compound 15: 1,2\alpha = 4, 2\alpha,2\beta = 15; 2\alpha,3 = 8; 3,4 = 10; 6\beta,7 = 7,8 = 7,7,11 = 13; 8,9\alpha = 6; 8,9\beta = 11; compound 15: 1,2\alpha = 4, 2\alpha,2\beta = 15; 2\alpha,3 = 8; 3,4 = 10; 6\beta,7 = 7,8 = 7,7,11 = 13; 8,9\alpha = 6; 8,9\beta = 11; compound 15: 1,2\alpha = 4, 2\alpha,2\beta = 15; 2\alpha,3 = 8; 3,4 = 10; 6\beta,7 = 7,8 = 7,7,11 = 13; 8,9\alpha = 6; 8,9\beta = 11; compound 15: 1,2\alpha = 4, 2\alpha,2\beta = 15; 2\alpha,3 = 8; 3,4 = 10; 6\beta,7 = 7,8 = 7,7,11 = 13; 8,9\alpha = 6; 8,9\beta = 11; compound 15: 1,2\alpha = 4, 2\alpha,2\beta = 15; 2\alpha,3 = 8; 3,4 = 10; 6\beta,7 = 7,8 = 7,11 = 13; 8,9\alpha = 6; 8,9\beta = 11; compound 15: 1,2\alpha = 4, 2\alpha,2\beta = 15; 2\alpha,3 = 8; 3,4 = 10; 6\beta,2 = 7,11 = 13; 8,9\alpha = 6; 8,9\beta = 11; compound 15: 1,2\alpha =$ $3\alpha_4 = 3$, $2\alpha_3\alpha_3 = 2\beta_3\beta_5 \sim 2$, $2\alpha_3\beta = 12$, $3\alpha_3\beta = 3\beta_4 = 12$, $6\alpha_7 = 75$, $6\beta_7 = 1$; $7\beta_8 = 75$, $7\beta_1 = 3.5$, $8\beta_8 = 10$; $8\beta_8 = 6$, compound 24: $1\alpha_1\beta = 1\alpha_2 = 3.3\alpha = 12$, $1\beta_2 = 2.3\beta = 5$, $3\alpha_4 = 6$, $3\beta,4 = 1.5, 4.15 = 7; 6.7 = 10; 9\alpha,9\beta = 16.5.$ × obscured.

Н	20	21	22*	23 (C ₆ D ₆)†	25	30 (C_6D_6) ‡	30 A¢§
1	5 91 br s	5 55 ddd	2.92 d	2.40 d	4 31 ddd	1 05 m	1.68 ddd
2α		} 5.75 ddd	3 37 dd	1 67 m	2 19 ddd	1 33 m	1 49 dddd
2β	_	3.73 aaa	} 3 3 / aa	1.34 m	1 49 m	0 96 m	1 77 m
3α	2 35 d	2.07 m	2 01 br dd	0.84 m	1.65	1 10 m	2 09 m
3β	2 41 dd	1 74 dddd	1 50 ddd	1 67 m	1 65 m	1 66 m	1 38 m
4	2.24 m	2 03 m	1 62 m	1 67 m	1 49 m	3.32 t	4 80 t
6α	1 90 dd	2 05 dd	1 95 dd	1 76 br d	1 77 ddd	1 79 dd	2 17 dd
6β	1 68 dd	1.96 br d	1.81 br d	2.56 d	2 02 ddd	1 01 dd	1 34 dd
7	2 98 ddddd	3 29 dddd	3 29 <i>dddd</i>	-	{ 2 44 ddd 2 36 ddd	2 16 ddddd	2 80 ddddd
8	4 62 ddd	5.04 ddd	5.09 ddd	4 64 br dd		3 69 ddd	4.22 ddd
9α	2 59 ddd	2 16 dd	2 30 dd	1 49 dd	6 10 <i>1</i>	0.90 q	$1\ 33\ q$
9β	2.88 dd	1.63 dd	1 67 ddd	1 64 dd	6 18 d	2 03 dt	2 36 dt
13	6.30 d	6 34 d	6 34 d	4 20 br dd		6 16 d	6 14 d
13'	5,67 d	5 61 d	5.60 d	4 13 br dd		4.97 d	5 41 d
14	1 09 s	0.74 s	0 70 s	0 42 s	1 11 s	0 55 s	095 s
15	0 96 d	0.93 d	0 82 d	0.50 d	093 d	0 62 d	0 96 d

Table 3. ¹H NMR spectral data of compounds 20-23, 25, 30 and 30Ac (400 MHz, CDCl₃, δ-values)

*OH 2 84 d †OH 2.01 t ‡H-10 1.18 m §H-10 1 80 m.

J [Hz]. Compound **20**: 1,9α = 1 5, 3α,3β = 17, 3β,4 = 5 5, 4,15 = 7, 6α,6β = 15; 6α,7 = 4.5; 6β,7 = 13, 7,8 = 8,9β = 7 5, 7,13 = 1.8; 7,13′ = 1 5; 8,9α = 9; compound **21**: 1,2 = 10, 1,3α = 1,3β = 2; 2,3α = 2, 2,3β = 4, 3α,3β = 13, 3β,4 = 10, 4,15 = 7, 6α,6β = 14; 6α,7 = 6, 7,8 = 9; 7,13 = 3 5, 7,13′ = 3; 8,9α = 7; 8,9β = 10, 9α,9β = 13; compound **22**: 1,2 = 4; 2,3β = 2 5, 3α,3β = 15, 3α,4 = 45; 3β,4 = 12; 4,15 = 7; 6α,6β = 14; 6α,7 = 7, 7,8 = 9; 7,13 = 3 5, 7,13′ = 3, 8,9α = 7; 8,9β = 10; 9β,ΟΗ = 15; 9α,9β = 13; compound **23**: 1,2α = 4, 4,15 = 7, 6α,6β = 13; 6α,13 = 8,13 ~ 1, 8,9α = 7, 8,9β = 11, 9α,9β = 13, compound **25**: 1,2α = 6, 1,2β = 12; 1,9 = 1.5; 4,15 = 7, 6α,6β = 6α,7β = 14, 6α,7α = 5, 6β,7α = 6β,7β = 4.5, 7α,7β = 16, compounds **30** and **30**Ac 3α,4 = 3β,4 = 9; 6α,6β = 14, 6α,7 = 6, 6β,7 = 12, 7,8 = 9, 7,13 = 3 5, 7,13′ = 3, 8,9α = 9α,9β = 9α,10 = 12; 8,9β = 9β,10 = 3 5, 10,14 = 7 (compound **30**Ac: 1,2α = 8, 1,2β = 1,10 ~ 10, 2α,2β = 2α,3β ~ 12, 2α,3α = 4 5)

EXPERIMENTAL

The air-dried aerial parts (700 g, voucher 88/48, deposited in the SWA Herbarium at Windhoek, collected near Okahandja in March 1988) were extracted and worked-up as reported previously [20] The extract was first separated by CC into six fractions. Fraction 1 gave by TLC 50 mg selina-4,11-diene and fraction 2 100 mg geranyl isobutyrate TLC of fraction 3 (Et₂O-petrol, 1 3) gave 60 mg 32 and 60 mg 33 HPLC (MeOH-H₂O, 4 1, always RP 8, ca 100 bar, flow rate 12 ml/min) of fraction 4 gave 450 mg 10 (R, 26 min) and 30 mg 11 $(R_t \, 30 \, \text{min})$. Medium pressure chromatography (MPC) of fraction 5 (silica gel ϕ 30-60 μ , Et₂O-petrol, 1 3 with increasing amounts of Et₂O) gave 45 fractions which were combined after monitoring by TLC into five fractions (5/1-5/5) Fraction 5/1 contained 100 mg 10, fraction 5/2 10 mg 11, fraction 5/4 100 mg 31 and 5/5 60 mg 30 Fraction 5/3 was separated by HPLC (MeOH-H₂O, 4·1) affording 12 mg 22 (R, 1 1 min), 10 mg 21 (R, 1.6 min) and a mixture (R_t 3.3 min) which gave by TLC $(Et_2O-petrol, 3:1)$ 5 mg **34** $(R_f \ 0.70)$ and 3 mg **19** $(R_f \ 0.65)$ Fraction 6 was also separated again by MPC (Et₂O-petrol, 1 3 with increasing amounts of Et₂O and finally Et₂O-MeOH, 9 1) The fractions obtained were combined into nine fractions (6/1-6/9). Fraction 6/1 gave 4 mg 27 and fraction 6/6 550 mg 5. HPLC of fraction 6/2 (MeOH-H₂O, 13 7) gave 15 mg 26 (R_t 3.6 min) and 60 mg 30 (R_t 5.1 min) HPLC of 6/3 (MeOH-H₂O, 13 7) afforded 10 mg **20** (R, 1.8 min), 8 mg **23** (R, 2.3 min) and 5 mg 25 (R, 26 min). TLC of fraction 6/4 gave 15 mg 1 and 15 mg 16 HPLC of 6/5 (MeOH- H_2O , 13.7) gave 6 mg 7 (R_1 42 min), 20 mg 16 (R_t 49 min), 2 mg 18 (R_t 6.0 min) and 1 2 mg 2 (R_t 7.0 min) HPLC of 6/7 (MeOH- H_2O , 3 2) gave 30 mg 9 (R, 45 min), $2 \text{ mg } 12 (R_t 28 \text{ min}) \text{ and } 2 \text{ mg } 3 (R_t 59 \text{ min}) \text{ HPLC of } 6/8$ (MeOH-H₂O, 3 2) afforded 4 mg 8 (R_t 2 4 min), 15 mg 14 (R_t 3 2 min) and three mixtures (6/8/1-6/8/3). Fraction 6/8/1 gave by TLC (CHCl₃-C₆H₆-Et₂O-MeOH, 30 30.30 1) 2 mg 15 (R_f 0 45) and 8 mg vomifoliol (R_f 0 20). Fraction 6/8/2 gave by TLC (same solvent) 3 mg 28 (R_f 0 58) and 2 mg 13 (R_f 0 42) TLC of 6/8/3 (Et₂O-MeOH, 30 1) gave a mixture of 4 and 6 After acetylation HPLC (MeOH-H₂O, 13 7) gave 5 mg 4Ac (R_t 6 6 min) and 10 mg 6Ac (R_t 6 9 min) HPLC of 6/9 (MeOH-H₂O, 1 1) gave 10 mg 24 (R_t 2 3 min) Known compounds were identified by comparing the 400 MHz ¹H NMR spectra with those of authentic material

3α-Hydroxy-11βH-eudesm-4-en-12,8β-olide (2) Colourless gum, IR $\nu_{\rm max}^{\rm CHCI_3}$ cm $^{-1}$ 3580 (OH), 1760 (γ-lactone), MS m/z (rel int.) 250.157 [M] $^+$ (47) (calc for C₁₅H₂₂O₃ 250 157), 235 (56), 232 (40), 161 (35), 143 (100), 119 (74), 105 (68), 91 (59), $[\alpha]_{\rm D}^{24}$ + 145° (CHCl₃, c 0 12)

3α-Hydroxy-11αH-eudesm-4-en-12,8β-olide (3) Colourless crystals, mp 123°; IR $v_{\rm max}^{\rm CHCl_3}$ cm $^{-1}$ 3580 (OH), 1765 (γ-lactone), MS m/z (rel int) 250 157 [M] $^+$ (58) (calc for C₁₅H₂₂O₃. 250 157), 235 (64), 231 (25), 217 (12), 189 (20), 161 (37), 143 (100), 124 (57), 119 (71), 105 (70), [α]_D²⁴ +99° (CHCl₃, c 0 54)

 2α -Hydroxy-eudesm-3,11(13)-duen-12,8 β -olude (4) Colourless gum, not free from 3 Acetylation (Ac₂O, 70°) gave 4Ac, colourless crystals, mp 153°, IR $\nu_{\rm max}^{\rm CCL}$ cm $^{-1}$ 1770 (γ -lactone), 1735, 1240 (OAc), MS m/z (rel int) 290 152 [M]⁺ (0 6) (calc for C₁₇H₂₂O₄ 290 152), 230 (36), 215 (23), 143 (94), 119 (100), 105 (63), 91 (60); [α]_D^{2 ξ^2} + 71° (CHCl₃, c 0 27)

 2α -Hydroxy-11α,13-dthydroalantolactone (6) Colourless crystals, mp 147°, IR $\nu_{\rm max}^{\rm CHCl_3}$ cm $^{-1}$ 3580 (OH), 1760 (γ -lactone), MS m/z (rel int) 250 [M] $^+$ (1), 232 146 [M – H₂O] $^+$ (100) (calc for C₁₅H₂₀O₂ 232 146), 217 (20), 159 (28), 143 (53), 119 (42), 105 (25) Acetylation gave 6Ac (Ac₂O, 70°) colourless crystals, mp 175°,

IR $v_{\text{max}}^{\text{CCL}}$ cm⁻¹: 1780 (γ -lactone), 1740, 1250 (OAc); MS m/z (rel. int.) 232.146 [M – HOAc]⁺ (76) (calc. for $C_{15}H_{20}O_2$: 232.146), 217 (21), 159 (37), 143 (200), 119 (48), 105 (32)

 2α -Hydroxy-11 β ,13-dihydroalantolactone (7). Colourless crystals, mp 134°, IR $\nu_{\rm max}^{\rm CHCl_3}$ cm $^{-1}$ · 3590 (OH), 1760 (γ -lactone); MS m/z (rel. int.): 250.157 [M] $^+$ (0.6) (calc. for C₁₅H₂₂O₃· 250.157), 232 (84), 217 (19), 159 (33), 143 (100), 119 (65), 105 (34), 91 (34); [α] $_{\rm D}^{\rm 24^o}$ + 9° (CHCl₃; c 0.18).

 $2\alpha,11\alpha$ -Dihydroxy-11,13-dihydroalantolactone (8). Colourless crystals, mp 186°; IR $\nu_{\rm max}^{\rm CHCl_3}$ cm $^{-1}$. 3580 (OH), 1765 (γ -lactone); MS m/z (rel. int): 266 152 [M] $^+$ (0.5) (calc. for $C_{15}H_{22}O_4$: 266.152), 248 (5), 160 (100), 145 (61), 119 (54), 105 (29), $[\alpha]_{\rm D}^{24^*}$ – 26° (CHCl $_3$, c 0.16).

1β,10β-Epoxyeremophil-11(13)-en-12,8β-olide (10). IR $v_{\rm max}^{\rm CCl_4}$ cm⁻¹· 1780 (γ-lactone); MS m/z (rel. int.): 248.141 [M] + (9) (calc for C₁₅H₂₀O₃: 248.141), 233 (12), 220 (12), 204 (8), 179 (31), 161 (37), 145 (32), 138 (57), 121 (51), 119 (44), 93 (72), 67 (77), 55 (100); [α]_D²⁴ - 17° (CHCl₃; c 3.68). Reaction of 10 with dil H₂SO₄ in dioxane afforded 16, identical with the natural product.

1 β ,10 β -Epoxy-11 α H-eremophilan-12,8 β -olide (11). Colourless crystals, mp 98°, IR $\nu_{\text{max}}^{\text{CCL}_1}$ cm⁻¹ 1780 (γ -lactone); MS m/z (rel. int.). 250.157 [M]⁺ (16) (calc. for C₁₅H₂₂O₃: 250.157), 235 (14), 182 (36), 181 (72), 133 (82), 119 (100), 93 (72), 91 (54).

 2α -Hydroxy-1 β ,10 β -epoxyeremophil-11(13)-en-12,8 β -olide (12) Colourless gum; IR ν ^{CCL4} cm ⁻¹: 3550 (OH), 1775 (γ -lactone); MS m/z (rel. int.). 264.136 [M] $^+$ (1) (calc. for C₁₃H₂₀O₄· 264.136), 246 (6), 208 (7), 181 (41), 119 (36), 84 (100).

 2α -Hydroxy-1 β ,10 β -epoxy-11 β H-eremophilan-12,8 β -olide (13). Colourless gum, IR $\nu_{max}^{CHCl_3}$ cm $^{-1}$. 3560 (OH), 1765 (γ -lactone); MS m/z (rel. int.) 266.152 [M] $^+$ (0.5) (calc. for $C_{15}H_{22}O_4$: 266.152), 183 (37), 153 (66), 84 (100).

 2α -Hydroxy-1 β ,10 β -epoxy-11 α H-eremophilan-12,8 β -olide (14). Colourless crystals, mp 145°, IR $\nu_{\max}^{\text{CHCl}_3}$ cm $^{-1}$: 3560 (OH), 1770 (y-lactone); MS m/z (rel. int): 266.152 [M] $^+$ (1) (calc for $C_{15}H_{22}O_4$: 266.152), 183 (12), 125 (37), 112 (100); $[\alpha]_D^{24^*} - 74^\circ$ (CHCl₃, c 0.4).

 3α -Hydroxy-1 β , $\overline{10}\beta$ -epoxy-11 β H-eremophilan-12,8 β -olide (15). Colourless gum; IR $\nu_{\text{max}}^{\text{CHCl}}$ cm⁻¹: 3400 (OH), 1760 (γ -lactone); MS m/z (rel. int.) 266 152 [M]⁺ (4) (calc. for $C_{15}H_{22}O_4$: 266.152); 251 (1), 248 (3), 233 (3), 181 (21), 121 (50), 111 (57), 55 (100).

1 β ,10 α -Dihydroxyeremophil-11(13)-en-12,8 β -olide (16) Colourless oil; IR ν $^{\rm CCl_4}_{\rm max}$ cm $^{-1}$ · 3600 (OH), 1775 (γ-lactone); MS m/z (rel. int.) 266.152 [M] $^+$ (12) (cálc for C $_{15}$ H $_{22}$ O $_4$: 266.152), 248 (10), 230 (20), 192 (21), 180 (66), 143 (32), 84 (100), 69 (53); [α] $_2^{\rm D4}$ $^+$ + 43 $^\circ$ (CHCl $_3$; c 1.14): NaBH $_4$ reduction gave 17, colourless crystals, mp 190 $^\circ$; IR ν $_{\rm max}^{\rm CHCl}$ $_3$ cm $^{-1}$: 3585 (OH), 1760 (γ-lactone); MS m/z (rel. int.): 268 168 [M] $^+$ (4) (calc. for C $_{15}$ H $_{24}$ O $_4$ · 268.168), 250 (6), 232 (12), 217 (5), 182 [M – MeCHCH $_2$ CH $_2$ CHOH] $^+$ (92), 109 [C $_7$ H $_9$ O] $^+$ (100).

 $1\beta,10\alpha\text{-}Dihydroxy\text{-}11\beta\text{H-}eremophilan\text{-}12,8}\beta\text{-}olide$ (18). Colourless gum; IR ν $_{\rm max}^{\rm CHCl_3}$ cm $^{-1}$: 3580 (OH), 1760 (y-lactone); MS m/z (rel. int): 268 168 [M] $^+$ (4) (calc. for C $_{15}{\rm H}_{24}{\rm O}_4$: 268.168), 250 (4), 232 (4), 182 (46), 129 (50), 109 (68), 84 (100).

10α-Chloro-1β-hydroxyeremophil-11(13)-en-12,8β-olide (19) Colourless crystals, mp 203°, IR $v_{\max}^{CHCl_3}$ cm $^{-1}$. 3580 (OH), 1760 (γ-lactone); MS m/z (rel. int.): 284.118 [M] $^+$ (2) (calc for $C_{15}H_{21}O_3Cl$ · 284.118), 266 [M $-H_2O$] $^+$ (1.5), 249 [M-Cl] $^+$ (3), 231 [249 $-H_2O$] $^+$ (5), 180 (100), 135 (21), 84 (48); [α] $_D^{24}$ $^+$ +41° (CHCl $_3$; c 0.46). Reaction of 10 in CHCl $_3$ with HCl at room temp. gave quantitatively the lactone 19.

2-Oxo-eremophila-1(10),11(13)-dien-12,8β-olide (20). Colourless oil; IR $\nu_{\rm max}^{\rm CCL_4}$ cm⁻¹: 1780 (γ-lactone), 1675 (C=CC=O); MS m/z (rel. int.) 246.126 [M]⁺ (16) (calc. for C₁₅H₁₈O₃: 246.126), 204 (26), 192 (30), 177 (52), 176 (70), 134 (50), 121 (74), 119 (83), 105 (100), 93 (96), 79 (91), 67 (64); CD (MeCN) $\Delta \varepsilon_{290} + 1.92$.

10α-Hydroxyeremophila-1,11(13)-dien-12,8β-olide (21). Colourless oil; IR $v_{max}^{\rm CCl4}$ cm⁻¹· 3620 (OH), 1775 (γ-lactone); MS m/z (rel int.). 248.141 [M]⁺ (calc. for C₁₅H₂₀O₃: 248.141), 233 [M - Me]⁺ (16), 230 [M - H₂O]⁺ (14), 215 [230 - Me]⁺ (84), 204 [M - CO₂]⁺ (82), 69 (100).

10α-Hydroxy-1α,2α-epoxyeremophil-11(13)-en-12,8β-olide (22). Colourless oil; IR $v_{\rm max}^{\rm CCl4}$ cm $^{-1}$ 3530 (OH), 1775 (γ-lactone), MS m/z (rel int.). 264.136 [M] $^+$ (6) (calc for C₁₅H₂₀O₄. 264.136), 246 [M $^-$ H₂O] $^+$ (8), 193 (88), 165 (40), 137 (54), 95 (81), 69 (100).

13-Hydroxy-1β,10β-epoxyeremophil-7(11)-en-12,8β-olide (23) Colourless crystals, mp 129°; IR $v_{\text{max}}^{\text{CCl}}$ cm $^{-1}$: 3400 (OH), 1775, 1755 (γ-lactone), MS m/z (rel. int): 264.136 [M]⁺ (9) (calc for $C_{15}H_{20}O_4$: 264.136), 249 [M – Me]⁺ (5), 246 [M – H_2O]⁺ (12), 126 [$C_6H_6O_3$]⁺ (68), 95 [126 – CH₂OH]⁺ (100); [α]₂²⁴ – 75° (CHCl₃; c 0.20).

 $2\alpha,10\beta$ -Dihydroxyondetianone (24). Colourless crystals, mp 148°, IR $\nu_{\rm max}^{\rm CHG_3}$ cm $^{-1}$ · 3600 (OH), 1675 (C=CC=O); MS m/z (rel. int.): 210.126 [M] $^+$ (1) (calc for ${\rm C}_{12}{\rm H}_{18}{\rm O}_3$ · 210.126), 192 (18), 177 (8), 124 (49), 112 (100), 110 (56), 96 (57), 82 (72), 67 (54), 55 (47), ${\rm [\alpha]_D^{24^\circ}+27^\circ}$ (CHCl $_3$, c 0.23)

1α-Hydroxyisoondetianone (25). Colourless oil; IR $v_{\text{max}}^{\text{CCI}}$ cm⁻¹: 3400 (OH), 1680 (C=CC=O); MS m/z (rel. int): 194 131 [M]⁺ (84) (calc. for C₁₂H₁₈O₂·194.131), 179 [M – Me]⁺ (38), 176 [M – H₂O]⁺ (11), 137 (74), 119 (64), 109 (100), 95 (96); [α]_D^{24°} + 140° (CHCl₃, c 0.27)

1β,10β-Epoxyeremophil-11(13)-en-12-oic acid (27). Colourless oil; IR $v_{\text{mfs}}^{\text{CCL}}$ cm⁻¹. 3500–2500, 1690, 1620 (C=CCO₂H); MS m/z (rel. int): 250 157 [M]+ (12) (calc. for C₁₅H₂₂O₃. 250 157), 235 [M-Me]+ (14), 191 (16), 152 (37), 147 (42), 121 (56), 105 (81), 91 (84), 77 (82), 55 (100); ¹H NMR (CDCl₃): δ 2 94 (d, H-1, J = 4 Hz), 1 97 and 1.71 (m, H-2), 1.32 and 1.17 (m, H-3), 1.70 (m, H-4), 1.82 (dd, H-6, J = 15, 5 Hz), 1.73 (dd, H-6', J = 15, 6.5 Hz), 3.02 (dt, H-7, J = 6, 6 Hz), 1.97 and 1 89 (m, H-8), 2.36 (m, H-9), 1 09 (dt, H-9', J = 14, 6 Hz), 6.30 and 5.74 (br s, H-13), 0.94 (s, H-14), 0.68 (d, H-15, J = 7 Hz).

2-Desoxy-4-epi-pulchellin (30) Colourless crystals, mp 135°; IR $v_{\text{max}}^{\text{CCl_4}}$ cm⁻¹: 3600 (OH), 1775 (γ -lactone); MS m/z (rel. int.): 250.157 [M]⁺ (22) (calc. for $C_{15}H_{22}O_3$: 250.157), 232 [M $-H_2O$]⁺ (28), 217 [232 -Me]⁺ (14), 149 (32), 136 (38), 121 (40), 109 (51), 108 (100), 81 (76); $[\alpha]_D^{124}$ + 43° (CHCl₃, c 0.59). Acetylation (Ac₂O, 1 hr, 70°) afforded 30Ac, colourless oil; IR $v_{\text{max}}^{\text{CCl_4}}$ cm⁻¹ 1775 (γ -lactone), 1740, 1250 (OAc); MS m/z (rel. int.): 292.167 [M]⁺ (1.3) (calc. for $C_{17}H_{24}O_4$: 292.167), 232 [M -HOAc]⁺ (24), 217 [232 -Me]⁺ (9), 108 (100).

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